

Supplementary information

**Table 1.** Analysis data for the different cyclic voltammograms of copper bromide ions in 0.1M NaOCl

M x10 <sup>-2</sup>	E <sub>pa</sub> Vol t	E <sub>pc</sub> Vol t x10 <sup>-1</sup>	ΔE <sub>p</sub> volt	(-) ip <sub>a</sub> x10 <sup>-4</sup>	ipc x10 <sup>-4</sup>	i <sub>pa</sub> /i <sub>pc</sub>	E <sup>0</sup>	D <sub>a</sub> x10 <sup>-11</sup>	D <sub>c</sub> x10 <sup>-11</sup>	E <sub>p<sub>c</sub></sub> /2	α <sub>na</sub>	K <sub>s</sub> x10 <sup>-4</sup>	Γ <sub>c</sub> x10 <sup>-7</sup>	Q <sub>c</sub> x10 <sup>-4</sup>	(-) Γ <sub>a</sub> x10 <sup>-7</sup>	(-) Q <sub>a</sub> x10 <sup>-4</sup>
1.43	0.1830	1.73	0.3560	3.07E	3.63	0.8456	0.0050	6.3611	8.90	0.0045	0.2633	7.21	1.2070	3.66	1.0207	3.09

**Table 2** Effect of concentration for ligand LC, 0.1 scan rate at 19.1<sup>o</sup>C (Cu<sup>+1</sup> ↔ Cu<sup>0</sup>).

L x10 <sup>-3</sup>	E <sub>pa</sub>	E <sub>pc</sub>	ΔE <sub>p</sub>	ipa X10 <sup>-4</sup>	ipc X10 <sup>-5</sup>	ipa/i <sub>pc</sub>	E <sub>0</sub>	D <sub>a</sub> X10 <sup>-11</sup>
0.04	0.0799	0.7877	0.7078	1.15	1.51	7.5980	0.4338	1.1677
0.09	0.0696	0.7723	0.7027	1.40	0.853	16.412	0.4210	1.7275
0.13	0.076	0.7823	0.7059	1.45	0.641	22.627	0.4293	1.8528
0.17	0.0735	0.7683	0.6948	1.43	0.653	21.840	0.4209	1.7947
0.22	0.0750	0.7448	0.6697	1.86	1.40	13.263	0.4099	3.0470
1.02	0.0698	0.2876	0.2178	0.842	2.32	3.6257	0.1787	0.6252
1.85	0.0817	0.0061	0.0878	0.707	4.98	1.4200	0.0378	0.4412
2.78	0.0875	0.0027	0.0848	0.231	4.88	0.4738	0.0451	0.005

**Table 3** Effect of concentration for ligand LC using 0.1 scan rate at 19.1<sup>o</sup>C (Cu<sup>+1</sup> ↔ Cu<sup>0</sup>)

Lx10 <sup>-3</sup>	DcX10 <sup>-13</sup>	Epc/2	α <sub>na</sub>	K <sub>s</sub> CX10 <sup>-9</sup>	Γ <sub>c</sub> X10 <sup>-9</sup>	(+) Q <sub>c</sub> X10 <sup>-</sup>	Γ <sub>a</sub> X10 <sup>-8</sup>	(-) Q <sub>a</sub> X10 <sup>-</sup>	Log β <sub>j</sub>	β <sub>j</sub>	(-)ΔG (KJ/mol)
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						<b>5</b>		<b>4</b>			
0.04	890	0.0045	0.2633	721	120.70	36.6	10.207	3.09	8.329	213791776	46.612
0.09	2.02	<b>0.082</b>	1.4076	2.06	5.0364	1.53	3.8266	1.16	8.120	131859127	45.438
0.13	0.641	0.7964	1.9463	1.43	2.8358	0.859	4.6544	1.41	8.275	188371398	46.304
0.17	0.362	0.8003	2.5921	1.20	2.1302	0.645	4.8202	1.46	8.140	138254984	45.553
0.22	0.376	0.7873	2.462	1.34	2.1721	0.658	4.7440	1.44	7.961	91534331	44.551
1.02	1.73	0.7787	1.3797	2.75	4.6605	1.41	6.1813	1.87E	4.160	14485.260	23.283
1.85	4.76	0.3410	0.8748	3.22	7.7227	2.34	2.8000	0.848	1.942	87.534	10.867
2.78	21.9	0.0685	0.6261	1220	16.564	5.02	2.3521	0.713	2.273	187.573	12.720

**Table 3** Analysis of the first wave of  $\text{CuBr}_2$  at different concentrations of the depolizer at  $19.1^\circ\text{C}$  for reaction, ( $\text{Cu}^{+2} \leftrightarrow \text{Cu}^{+1}$ ).

$v$	E <sub>Pa</sub>	E <sub>pc</sub>	$\Delta E_p$	(-) I <sub>p,a</sub> $\times 10^{-5}$	I <sub>p,c</sub> $\times 10^{-5}$	ip <sub>a</sub> /ip <sub>c</sub>	E <sub>o</sub>	D <sub>a</sub> $\times 10^{-10}$	D <sub>c</sub> $\times 10^{-11}$	E <sub>pc</sub> /2	$\alpha_{\text{nac}}$	K <sub>s</sub> $\times 10^{-5}$	$\Gamma_c$ $\times 10^{-8}$	(+) Q <sub>c</sub> $\times 10^{-4}$	(-) Q <sub>a</sub> $\times 10^{-4}$
0.1	0.5899	0.0034	0.5864	1.41	2.52	0.5608	0.296	0.013	0.042	0.0534	52.620	6.95	0.8366	0.25	0.142
0.05	0.0371	0.00777	0.0449	10.3	2.19	4.6871	0.014	0.142	0.064	0.0547	0.2431	0.19	1.4544	0.44	2.07
0.02	0.0516	0.0082	0.0598	12.9	5.92	2.1707	0.021	0.558	1.184	0.0525	3.0905	2.13	9.849	2.98	6.48
0.01	0.0715	0.0329	0.0385	13.6	9.74	1.3997	0.052	1.25	6.400	0.0765	1.0611	1.66	32.377	9.81	13.7